The Slow Decline and the Subsequent Rise of Chlorophyll Fluorescence Transients in Intact Algal Cells

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ABSTRACT

The slow decline and the subsequent rise of chlorophyll fluorescence transients in intact algal cells. Mohanty, P. and Govindjee (Dept. of Botany, University of Illinois, Urbana, Illinois 61801, U.S.A.). Plant Biochemical Journal 1(2)78-106, 1974. Dark-adapted algae show changes in chlorophyll a fluorescence yield with time of illumination (the Kautsky effect) and there are three rapidly occurring sequential phases in the fluorescence transients. The effect of light, temperature, preillumination, inhibitors of electron flow, anaerobiosis, election acceptors, uncouplers and cofactors of photophosphorylation on the peak (P) to quasi-steady state level (S) fluorescence decay and S to second maximum level (M) fluorescence rise has been compared. On the basis of the data presented, together with the authors' previously published results, it is concluded that the P to S decay, although not directly linked to the oxidoreduction of carriers of electron transport chain, is related to the energy conservation process associated with the non-cyclic electron transport and photosynthetic oxygen evolution. The S to M rise does not need a sustained noncyclic electron flow and is perhaps related to the export of energy from, and import of ions to, the chloroplasts. Both the P to S and S to M changes are suggested to occur through associated microstructural changes of membrane matrix that are caused by ionic flow leading to the changes in the conformation of key protein-chlorophyll complexes.

Index words: Anacystis | Chlorella | Porphyridium | Chlorophyll fluorescence transients | conformational changes in membrane structure.

INTRODUCTION

When a dark adapted aerobic sample of intact algal cells or a leaf-segment is illuminated with saturating intensity of light, the fluorescence of chlorophyll a

¹ Abbreviations: CCCP, carbonyl cyanide 3-chlorophenylhydrazone; Chl a, chlorophyll a; DCMU, 3-(3, 4 dichlorophenyl)1, 1-dimethyl urea; FCCP, carbonyl cyanide 4-trifluormethoxyphenyl hydrazone; MV, methyl viologen; PBQ, p-benzoquinone; PMS, N-methyl phenazonium methyl sulphate; PS I (II), pigment system I (II); Q, primary electron acceptor of PS II; Tris, tris (hydroxymethyl) amino methane.

(Chl a) shows characteristic oscillations (see reference 1 for a review). cense transient begins from a low origin (0) followed by three rapidly occurring sequential phases within about one second of illumination: an inflection (I), a dip (D), and a peak (P). From P, the fluorescence yield shows a slow decline to a quasi-steady state level (S). After reaching the S level the fluorescence yield slowly rises to a second maximum (M) level and then declines to a level (T) within a few minutes of illumination. The T level represents a terminal level as no significant change in the fluorescence yield occurs thereafter. The fast 0→I rise has been suggested by Joliot (2) to be related to the changes in the redox states of the photoenzyme "S" involved in O₂ evolution (see reference 3). Thus the $0 \rightarrow I$ rise would reflect the final state change in the charge accumulation process leading from S₃ to S₄ state. Thus, both fluorescence and O₂ evolution rise in parallel during this The subsequent rise to P in the fluorescence transient of Ch1 a is believed to be due to an accumulation of the most of the electron transport carriers of photosynthesis in the reduced state (4). The D->P rise in the fluorescence transient is accompanied by a complementary decline of oxygen evolution (5-8) as would be expected if Chl a fluorescence were to compete with the photochemical reactions leading to oxygen evolution. The I -> D phase has been suggested to be partly due to the interaction of pigment system I (PS I) with PS II (4,9). Although the P to S decline phase of the fluorescence transient exhibits an antiparallel behaviour with that of oxygen evolution, it does not reflect a simple reoxidation of the primary electron acceptor Q of PS II. This puzzling behaviour was first documented by Duysens and Sweers (10) who observed that if the exciting light was turned off at the S level and then quickly turned on again, the yield of fluorescence remained at the low S level and did not rise up to the P level unless the dark period was long (minutes).

Duysens and Sweers (10) suggested that during the P to S decline phase the primary acceptor of PS II (Q) undergoes a change to a form denoted as Q' which has a reduced ability to accept electrons but it still quenches fluorescence like Q. In recent years Duysens (11) has modified this hypothesis of the transformation of quencher to accommodate new experimental data (12, 13; see also refrence 1). It has now been suggested that the P to S decline occurs, to a large extent, due to some sort of transformation of "physiological states" of thylakoid membranes facilitating flow of excitation energy from PS II to PS I (see reference 14 for a review) or due to alteration of deexcitation process within (PS II) units (see 7, 15, 16). Furthermore, Papageorgiou (17) has recently suggested that an ultrastructural change of the thylakoid membrane, due to the acidification of the thylakoid interior, facilitates

diffusion of quenchers notably molecular oxygen. In support of this hypothesis is the observation that the P to S decline is absent in anaerobic cells (4, 9, 18).

Krause (19) working with the intact leaves has shown that slow P to S type fluorescence quenching upon prolonged illumination accompanies an increase in absorbance at 540 nm due to shrinkage of chloroplast structure. He has shown that this type of Chl a fluorescence lowering depends not on the redox state of the quencher (Q) but on the high energy state of photophosphorylation. and Barber et al. (21) have also shown that P to S decline in the fluorescence yield reflects the efflux of cations like Mg²⁺ from grana to stroma region and back influx of the cations into the grana stacks in the subsequent darkness. As cations and proton movements are energy-linked processes, both the hypotheses of Papageorgiou (17) and of Krause (19, 20) would implicate a high energy state to be associated with the fluorescence decline from P to S. Cofactors like PMS (phenanzine methosulphate) appear to cause a P to S type lowering of Chl a fluorescence in isolated broken chloroplasts and in algae in the presence of DCMU (22-24). This observation would indicate that some kind of cyclic photophosphorylation may also be involved in P to S decline of Chl a fluorescence. Slovacek and Bannister (16) have shown that depletion of CO₂, (a condition which would be expected to induce cyclic photophosphorylation in vivo,) does not alter P to S decline of the fluorescence transient in Chlorella, althouth S to M rise phase is altered.

In view of these recent developments, we considered it desirable to reinvestigate the P to S decline part of the fast fluorescence transient in three commonly used algae—the green alga Chlorella, the red alga Porphyridium and the blue-green alga Anacystis. The effects of various treatments including inhibitors of photosynthesis and uncouplers of photophosphorylation on the P to S decline and on the subsequent S to M rise of the fluorescence transient are presented here. The lack of correspondence between the oxidation level of Q (10) and the P to S decline and at the same time the sensitivity of this phase of induction to a variety of uncouplers (see 1,17) indicate that the slow P to S decline may be linked to the development of "energy pressure" associated with photophosphorylation. Similarly, it has been shown that the S to M rise is not directly linked to photochemical reactions, as it is evident from the loss of complementarity between the kinetics of O₂ evolution and fluorescence changes during this period (25). The S to M rise is partly insensitive to inhibitors and partly sensitive to uncouplers; it exhibits large alterations in the emission characteristics of Chl a fluorescence which is indicative of phototransformation of 'states' of thylakoid membranes (see 1,14). These and other results, along with the observations presented here, suggest that the most part of the slow fluorescence change is not linked to the photochemical reactions, but is related to

structural changes of the membrane that may be associated with export and import processes between chloroplasts and cytoplasmic constituents.

MATERIALS AND METHODS

Intact cells of *Chlorella pyrenoidosa*, *Anacystis nidulans* and *Porphyridium* cruentum were used. Cells were grown in continuous light as described by Govindjee and Rabinowitch (26).

Cells were suspended in fresh growth medium or in phosphate or Tris buffer. Fluorescence was measured with a spectrofluorometer described by Shimony et al. (27). For the measurement of fluorescence transient, the procedures of Munday and Govindjee (4) and Mohanty et al. (8) were followed. All the chemical additions were made at least 10 minutes before measurements. Oxygen measurement was made as described by Mohanty et al. (8).

RESULTS

It is now well known that the fast OIDPS transient is seen in almost all types of intact cells including those of blue green algae (28). Compared to other algal cells, the OIDPS transient in blue green algae is somewhat lower in magnitude than the SMT transient. Fig. 1 shows the IDPS transient in two different cultures of Anacystis; the magnitude of P to S decline depends very much on the culture conditions. Munday (29) observed similar dependence of P to S decline on the culture conditions in Chlorella. Fig. 2 shows both the fast (IDPS) and the slow SM transients in another unicellular blue-green alga Microcystis. It has been shown earlier by Papageorgiou and Govindjee (25) that blue green algae exhibit a pronounced S to M rise and a very slow M to T decline. The S to M rise is resistant to DCMU treatment in Anacystis (see 24, 28).

1. Intensity dependence

In *Porphyridium*, D to P rise and P to S decline portions of the fast transient saturate at different light intensities (8). Fig. 3 shows a plot of the amplitudes of O, P and S as function of exciting light intensities in *Porphyridium*. Except for the level at O, the fluorescence intensities, at P and at S, do not show linear rise with increase in intensity of light. As the slope of the fluorescence versus intensity curve is proportional to the quantum yield of fluorescence, the yield at O does not change with intensity as shown earlier by Lavorel (30) and Govindjee *et al*. (31). The plot of relative yields at P and S represented as (P-O)/O and (S-O)/O, indicates that the variable yield at S tends to saturate at much lower intensity than

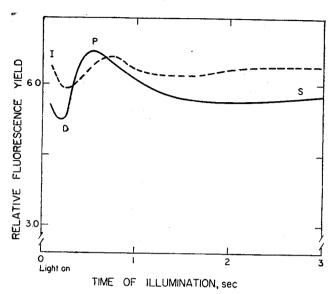


Fig. 1. Time course of Chl a fluorescence of two different cultures of Anacystis. λ excitation, 633 nm; intensity, 26 K ergs cm⁻² sec⁻¹; λ observation, 685 nm; C.S. 2-61 before the analyzing monoehromator. Dashed line, cells grown in high light (100 watts, 13 cm away); solid line, cells grown in low light (40 watts, 13 cm away); 6-day-old samples suspended in growth medium; both samples were adjusted at the "red" absorption peak of Chl a to 0.3 optical density for 1 cm path length; OI rise not recorded.

at P. Fig. 4 is a similar graph showing the intensity dependence of the variable fluorescence yield at P, (P-O)/O, and at S, (P-S)/S, in *Chlorella*. Here again, the variable fluorescence at S is saturated at relatively lower intensity than at P. This observation suggests that the P to S decline is not directly related to the development of the D to P phase; the latter has been shown to be due to accumulation of reduced carriers (9).

2. Temperature dependence

Rabinowitch (32) has reviewed early observations on the temperature dependence of fluorescence transients. Franck *et al.* (33) have shown that the P to S decline is inhibited at low temperatures. Fig. 5 shows the IDPS transient measured at 5° and 23°C in *Porphyridium*. Here we observe that both D to P rise and the

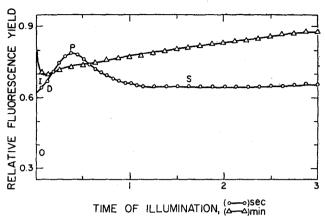
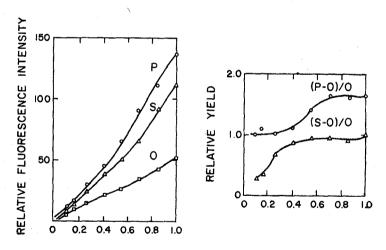


Fig. 2. Time-course of Chl a fluorescence transient in Microcystis. λ excitation, 633 nm; λ observation, 685 nm (half band width, 6.6 nm); C.S. 2-58 before the analyzing monochromator; open circles, fast transient; open triangles, slow transient; 6-day-old cultures used. Note that the abscissa has two time scales; 15 min dark period between each measurement; the fast OI not recorded.



RELATIVE EXCITAION INTENSITY

Fig. 3. The amplitude of Chl a fluorescence yields at 0, P and S (left) and the relative variable yield (right) versus excitation intensity in *Porphyridium*. Intensity 1.00=14 Kergs cm⁻² sec⁻¹. Relative yields were expressed as (P-O)/O or (S-O)/O; 8-day-old cells suspended in carbonate-bicarbonate and NaCl buffer (pH 8.5).

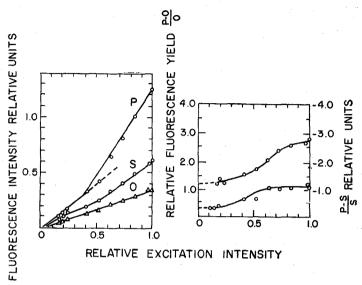


Fig. 4. Relative variable fluorescence yield at P represented as (P-O)/O (right, upper curve) and at S as (P-S)/S (right, lower curve) in *Chlorella* as function of excitation intensity. Left: fluorescence intensities at O, P, and S as function of exciting light intensity. Intencity 1.00=12 K ergs cm⁻² sec⁻¹. Excitation blue light, white light filtered through C.S. 4-72 and C.S. 3-73 filters.

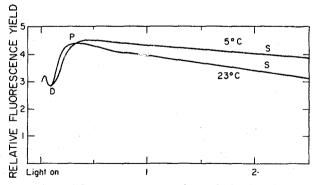


Fig. 5. The IDPS fluorescence transient of *Porphyridium* incubated at two different temperatures during dark adaptation; other details as in Fig. 3.

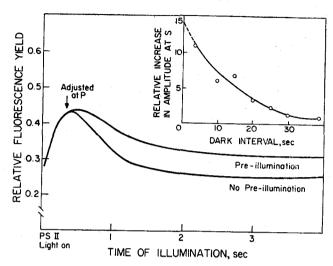


Fig. 6. The effect of 60 second system I preillumination on the decline in the fluorescence transient of *Porphyridium*.

Preillumination, 435 nm light (half band width, 10 nm); intentisy, 0.8 K ergs cm⁻² sec⁻¹. Insert: dark decay of the enhancement of S level by PS I preillumination.

P to S decline are slowed down at 5°C; the P to S decline seems to be more affected than the rise from D to P. If the slower P to S decline at 5° C, as compared to the one measured at 23° C, indicates a slower reoxidation of reduced Q by the pool of intersystem carriers, a faster D to P rise should have been observed, as this rise is associated with the accumulation of reduced Q (QH) (see 4, 9, 10). But this does not seem to be the case (Fig. 5). It has been shown (see 6, 7, 8), that the rise in the rate of O₂ evolution after the initial O₂ 'gush' (spike) and the slow P to S decline of Chl a fluorescence are antiparallel. This antiparallel (or complementary) behaviour of O₂ evolution and Chl a fluorescence would indicate that the Chl a fluorescence during this induction (P to S decline) period is linked to the redox state of Q, the quencher of fluoescence. We have observed, however, that the rate of rise of O₂ evolution after the initial O₂ 'gush' does not differ appreciably when measured at 5° C and at 25° C, although the P to S decline is significantly altered at these two temperatures (see fig. 5). This again suggests that P to S decline of Chl a fluorescence is not solely governed by the oxidoreduction state of Q. It is also clear from Fig. 5 that P to S decline is not "affected" by the extent of D to P rise.

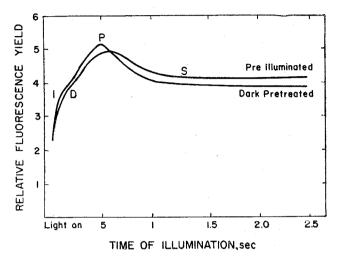


Fig. 7. Fluorescence induction in Chlorella cells. Note enhancement of the amplitude of S level fluorescence by system I preillumination. Dark time, 7 min; preillumination, 710 nm light $(1.3 \times 10^2 \text{ ergs cm}^{-2} \text{ sec}^{-1})$ for 60 seconds. Transient was measured 2 seconds after preillumination. Data obtained at room temperature.

3. Preillumination effects

Munday and Govindjee (9) have shown that P to S decline and the final S level in Chlorella is influenced by the wavelength of the preilluminating light. A preillumination of the sample with light absorbed mainly by PS I slowed down the P to S decline and raised the level of S (9, 28, 31). Fig. 6 shows similar results on the effect of preillumination for one minute with 435 nm light (mainly PS I) on the P to S decline in *Porphyridium*. (In this figure we have adjusted the fluorescence yield at P to read the same value as we assume that at P almost all the Q are reduced to QH.) The half-time for the decay of the preillumination effect was found to be approximately 12-15 seconds. Fig. 7 shows a similar transient obtained with Chlorella. Although a two-second dark-time intervened between preillumination (PS I) and excitation (PS II) light, it is nevertheless clear that the PS I-light preillumination slowed down both the D to P rise and the P to S decline. The slower D to P rise must be due to the accumulation of the intersystem carriers (A pool) in an oxidized state by the preilluminating PS I light, but the slower P to S decline cannot be explained by the oxidation state of the electron transport carriers (9, 29, 34).

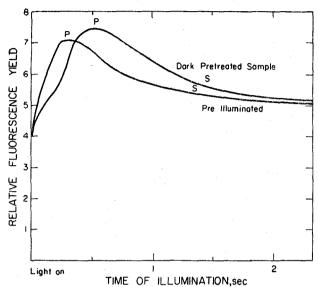


Fig. 8. Effect of brief system II preillumination on the Chl a fluorescence transient in Chlorella. Dark pretreatment, 7 min; preillumination with blue-green exciting light until P was reached, the light was turned off, and, then turned on again. The two transients are adjusted at "O". Cells were suspended in 0.05 M phosphate buffer (pH 8.0) plus 0.01 M NaCl. Intensity of exciting blue light, 14 K ergs cm⁻² sec⁻¹.

Fig. 8 shows fluorescence transients from a dark adapted sample and a sample of *Chlorella* preilluminated with predominantly PS II light. In the latter case the light was on until the P level was reached and turned off for 2 sec, and then the same light was turned on again and the complete transient was recorded. The short term preillumination with PS II light caused, as expected, a rapid D to P rise owing to an accumulation of reduced "A" pool caused by the PS II preilluminating light. However, the faster P to S decline in the preilluminating sample, as compared to the dark-adapted one, cannot be easily explained on the basis of a simple reoxidation of electron carriers.

In summary, the PS I preillumination retards the P to S decay while PS II preillumination hastens it. This behaviour is contrary to what one would expect from the known antagonistic effect of PS II and PS I light on the redox state of Q. We also note that in *Chlorella*, there is no quenching of fluorescence by PS I

light at the S level, and the extent of quenching brought about by PS I light during the S to M rise portion of the transient is less than during the subsequent M to T decline phase (35).

4. Decay of P to S in light or in dark

Heath (36) reported that the dark decay of maximal fluorescence (F_{max} in isolated chloroplasts, without any exogenous acceptor) is very complex and the yield falls to the initial level only after 1 to 2 min. of darkness. The decay of P level in *Porphyridium* was found to be much faster than in broken isolated chloroplasts. Fig. 9 shows that in *Porphyridium* the yield at P declines much faster in

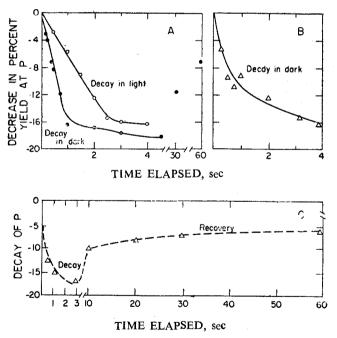


Fig. 9. The dark decline of the P level of Chl a fluorescence in Porphyridium. After dark adaptation, the system II light was turned on until P level was reached and then it was turned off. After dark interval, as shown on the X axis, the light was turned on again and the complete transient was recorded. The decrease in the amplitude of P by the second illumination is plotted as percentage decay. A and B, dark decay of P from two different cultures; for comparison, the P to S decay in light is also plotted as percented decay with time in A. C, rapid decay and slow restoration of P in darkness.

darkness than in light $(t_{1/2} = 0.25\text{-}0.5 \text{ sec} \text{ in dark}; 1.5 \text{ sec} \text{ in light})$. One may expect the oxidation of QH to be faster in light than in dark as part of the exciting light is absorbed by PS I. However, the results shown in fig. 9 are contrary to this expectation. Again, although fluorescence decays to a low level fairly rapidly in dark (1-3 sec), yet the restoration of P level in subsequent illumination can be achieved only after longer (>10 sec) dark adaptation (fig. 9c). These results clearly show that P to S decline does not reflect the redox state of Q and other processes regulate the fluorescence yield during this period of induction.

5. Effects of inhibitors

While the preceding observations indicate that the P to S decline does not seem to directly reflect the oxidation of reduced primary acceptor (OH), nevertheless, the photosynthetic noncyclic electron flow may be thought to be necessary for the P to S decline. This is because of the fact that DCMU abolishes this decline. It is known that DCMU blocks the electron flow very close to Q, inhibiting the flow of electrons from Q to the pool of intersystem intermediates (A) (37). Unlike DCMU, hydroxylamine (NH₂OH) has been shown to inhibit oxygen evolution but not the electron transport between two photosystems (see 38,39). DCMU accelerates the D to P rise and NH₂OH slows it down. Fig. 10 shows that like DCMU, NH₂OH, at a high concentration (10mM), also inhibits the characteristic P to S decay in Porphyridium. Mohanty et al. (39) have shown that at this concentration NH₂OH feeds electrons to photosystem II and a noncyclic electron flow exists between the two photosystems. It is likely that, with NH₂OH as electron donor for PS II, the photophosphorylation ability of cells is reduced and this may have indirectly caused the suppression of P to S decay. would suggest that P to S decline is indeed complex. [In Anacystis, unlike the case in Porphyridium, both NH₂OH and DCMU together do not suppress S to M rise but abolish P to S decline (28).]

6. Effect of anaerobiosis

Kautsky et al. (18) reported that under anaerobic conditions there is no P to S decline. Munday and Govindjee (4) showed that in *Chlorella* under anaerobic condition the P to S decline was abolished, although the IDP phase of the fast transient was quite prominent under this condition. Franck and Hoffman (40) and Schreiber et al. (41) have also reported a progressive inhibition of the P to S decline with increasing anaerobiosis. One may expect anaerobiosis as well as DCMU treatment to induce a cyclic flow of electrons that supports photophosphorylation (42). If this view is correct, the prevention of P to S decline by anaerobio-

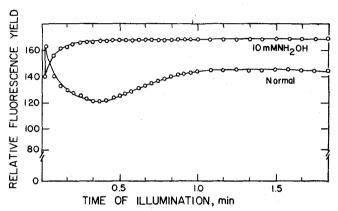


Fig. 10. The slow fluorescence transient in normal and NH₂OH (10 mM)-treated *Porphyridium*.

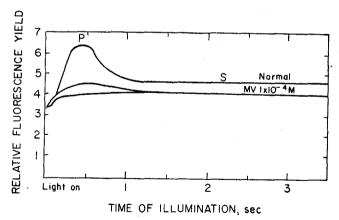
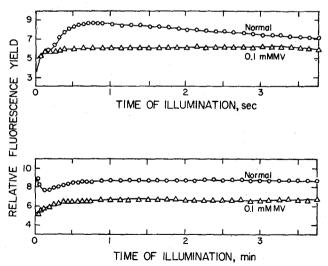


Fig. 11. Effect of methyl viologen on the DPS transient in Chlorella. Cells suspended in carbonate-bicarbonate buffer (0.1 M, pH 9.2). Upper trace, untreated sample; middle trace, 1 minute after addition of 0.1 mM MV; lower trace, 10 min. after the addition of MV. In each case sample was kept 7 min in darkness before measurement.

sis or by DCMU poisoning would indicate that the cyclic electron flow is not directly linked to P to S decline. Papageorgiou et al. (43) showed that the slow decline of Chl a induced by reduced PMS (PMSH₂) in DCMU-treated chloroplasts is also susceptible to oxygen tension suggesting that the P to S decline may include quenching by molecular oxygen.

7. Effect of system I electron acceptor (methyl viologen)

Munday and Govindjee (9) showed that methyl-viologen (MV) quenches the fast (DP) Chl a fluorescence rise in Chlorella. Mohanty (44) confirmed these results in Chlorella, Porphyridium and Anacystis. Fig. 11 shows that in the presence of 0.1 mM MV, the OI phase was not affected but MV suppressed the development of P and lowered the S level in Chlorella. Fig 12 shows our results obtained with Anacystis cells. We have observed that 0.1 mM MV gradually inhibits O₂ evolution with time of incubation in dark and finally O2 uptake is observed in lieu of O2 evolution. Also, in Chlorella the extent (amplitude) of cytochrome f oxidation by system I light was found to be roughly 2.5 times greater in MV-treated samples than in the These results indicate that MV accelerates the flow of electrons untreated samples. in the intact algal cells as it does in isolated chloroplasts. Thus, the suggestion of Munday and Govindjee (4,9) that the D to P rise is brought about by the transitory



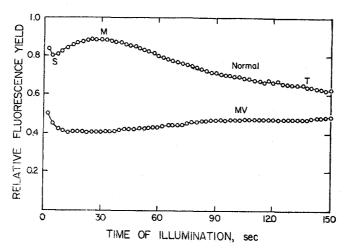


Fig. 13. The slow transient in Chl a fluorescence yield with and without Methyl viologen (MV) in Chlorella. Upper curve, untreated cells; bottom curve, with 1 mM MV. Fluorescence was measured at 685 nm; λ excitation, blue light (C,S. 4-72+C.S. 3-73); intensity, ≈10 K ergs cm⁻² sec⁻¹.

accumulation of most of the electron transport carriers in reduced state seems to be consistent with our observations. As noted above, MV eliminates most of the D to P rise and lowers the S level; this may indicate that an accumulation of reduced Q is a prerequisite for and is linked to the P to S decay, although the latter does not merely reflect the oxidation of QH.

Besides the effect on ODPS transient, MV also arrests most of the development of the slow SMT type of fluorescence change in *Chlorella* (see fig. 13). We do not know the reason for the absence of the S to M rise. This may be due to the fact that MV in light might have caused an 'over oxidation' of carriers and, thereby, perhaps imposed some irreversible alteration of structure. It may be added that we could not remove the effect of MV from the treated cells by washing the samples with buffer. MV, thus, in intact cells also may bring about some irreversible structural change.

8. Effect of system II acceptor (quinone)

Figure 14 illustrates the alteration of the time course of fluorescence in *Chlorella* by 0.1 mM benzoquinone (PBQ) a well known Hill oxidant (see 45).

PBQ suppressed the P level and lowered the amplitude of S (PBQ also affected slightly the O level, not shown). Since PBQ acts as a Hill oxidant for algal cells, we conclude that the suppression of the variable fluorescence (D to P phase) is due to the oxidation of QH by quinones. The slow rise that persists the quinone treatment is reminiscent of the S to M rise although this rise was slower here than in the normal cells. We note that a slow rise of fluorescence yield was invariably seen with PBQ-(fig. 14) but not with MV-treated *Chlorella* (fig. 13).

We do not know why the slow S to M rise in Chl a fluorescence is resistant

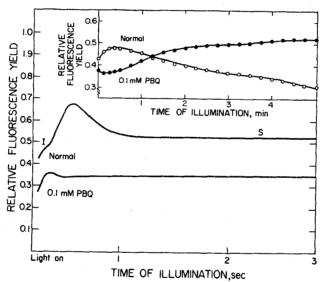


Fig. 14. Time course of the relative Chl a fluorescence yield in Chlorella with and without p-benzoquinone (PBQ). Main figure shows the fast transient, and the insert the slow transient.

to PBQ but not to MV. It may be possible that cyclic photophosphorylation is suppressed in the presence of MV, but not in the presence of PBQ. This would suggest that the S to M fluorescence rise is linked to cyclic electron flow-mediated phosphorylation. However, one difficulty in assuming this is the fact that no alteration in Chl a fluorescence yield is observed in Chlorella in the presence of DCMU (44,46). The slow and the persistent S to M rise observed in the presence of PBQ may be related to the change in the so-called "constant" yield fluorescence.

Amesz and Fork (47) have shown that the quenching of Chl a fluorescence by quinones may be of two kinds: (i) a dynamic physical quenching similar to what has been seen in case of Chl a solution and (ii) quenching due to oxidation of QH. Our results indicate that the main cause of lowering of DP transient by PBQ is due to its action as a Hill reagent. However, it appears that the slow S to M rise is not at all linked to oxidation state of Q; also, this slow rise does not seem to depend on the development of the fast (OIDPS) transient.

9. Effect of uncouplers of phosphorylation

The effect of uncouplers of photophosphorylation on the fluorescence yield changes has been reported by many investigators (see 1). One of the most studied uncoupler is carbonyleyanide hydrazone (CCCP or FCCP). We have recently reported the effect of CCCP on the IDPS transient of Porphyridium and Chlorella (24). FCCP at a relatively high concentration (10-30 μ M) was found to suppress The P to S ratio (measured after 3 sec of illumination) was the development of P. Although FCCP lowered the P to S decline, the slow S to M type also lowered. rise of Chl a fluorescence was not completely inhibited by this uncoupler. shows the slow S to M rise of Chl a fluorescence in Chlorella in the presence of 10 µM FCCP. Note that the rise of fluorescence from low S level is very slow and The steady state level of fluorescence is also high. there is no M to T decline This slow fluorescence rise in the presence of FCCP was significantly more at a lower pH (6.0) of the suspending medium than when the pH was high (8.0). This supports the earlier observation of Papageorgiou and Govindjee (46) that FCCP stimulates slow S to M rise in Chlorella although it suppresses both the P to S and the M to T decline in this alga.

Fig. 16 (top) shows traces of the fast fluorescence transient of *Anacystis* with and without CCCP. At a low concentration of CCCP (5 μ M) the amplitude of P is low and the P to S decline is also lowered, but the slow fluorescence rise is not altered at this concentration. However, with 10 μ M of FCCP, the P level was further depressed and the S to M rise was also arrested (also see 25, 28, 46). Fig. 16 (bottom) shows the effect of 5 and 10 μ M CCCP on the slow fluorescence changes in DCMU-poisoned *Anacystis* cells. 10 μ M CCCP did not cause the same extent of suppression of S to M rise in the presence of DCMU as in its absence. Papageorgiou and Govindjee (46), however, have observed a more dramatic differential effect with FCCP in normal and DCMU-treated *Anacystis* cells.

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Fig. 17 shows the effect of the uncoupler FCCP (5 μ M) on the rate of O₂ evolution in *Anasystis*. FCCP depressed the O₂ spike and caused a transitory O₂

uptake. With time, the rate of O_2 evolution increased although at a slower rate than in the control. It seems FCCP accelerated O_2 uptake in these cells and slightly suppressed the net O_2 production. This inhibition of O_2 production may be either

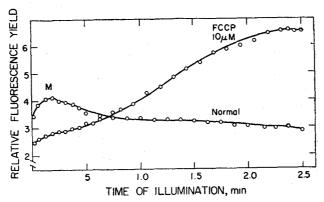


Fig. 15. Time course of slow changes in Chl a fluorescence yield in Chlorella with and whithout 10 µM FCCP. Buffer, phosphate and NaCl.

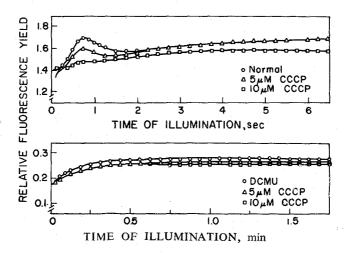


Fig. 16. Effect of the uncoupler of phosphorylation (CCCP) on the fluorescence yield of Chl a in normal and DCMU poisoned Anacystis. Top figure, fast fluorescence changes in normal cells. Bottom figure, slow fluorescence changes with 15 \(\mu \)M DCMU added. Buffer, 50 mM Tris-Cl plus 1 mM NaCl, pH 7.8.

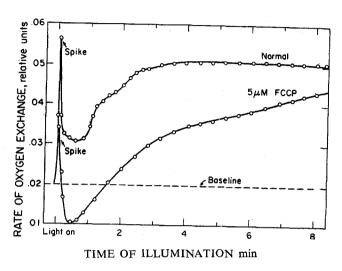


Fig. 17. Time course of O₂ exchange in *Anacystis* in the presence of FCCP.

due to the uncoupling of photophosphorylation or to the photoinduced O₂ uptake. Unlike CCCP, NH₄Cl did not seem to alter the characteristic pattern of OIDPS (48) in *Chlorella*. In some cultures, however, we observed a depression of the P to S decline; in others, it remained unaltered.

Slovacek and Bannister (49) have shown that NH₄Cl enhances the slow S to M rise in the *Chlorella* depleted of CO₂ supply. These cells show no S to M rise in the absence of NH₄Cl although one would expect that these CO₂ depleted cells of *Chlorella*, like normal cells, possess the ability for cyclic photophosphorylation. Hence, the enhancement of S to M rise by the addition of NH₄Cl to these CO₂-starved cells cannot be related to the uncoupling effect of this chemical.

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Slovacek (unpublished results) has also observed that depletion of CO₂ increases the ratio of System I to System II fluorescence and NH₄Cl reverses this effect and causes an increase in PS II fluorescence. This is in confirmity with the results of Mohanty et al. (8) that in the S to M rise the M level exhibits more PS II fluorescence than the S level. Unlike the case in Chlorella, the S to M rise in Anacystis is suppressed by uncouplers, notably salicylanalides and FCCP; these uncouplers, also, arrest the macroscopic structural alteration as measured by change in optical density at 540 nm (50). Furthermore, the glutaraldehyde fixed cells do

not show S to M fluorescence rise (28,39). This suggests that the S to M rise of fluorescence may be linked to some structural alterations of the thylakoid membranes.

10. Effect of cyclic electron transport confactor (PMS)

Phenanzine methosulfate (PMS) is known to accelerate cyclic flow in isolated chloroplasts. In intact cells of *Anacystis*, reduced PMS (PMSH₂) abolishes the P to S decline (24). However, the fluorescence yield slowly increases from a low level to a level approximately that of the P level and then remains unaltered. There is no P to S decline, although fluorescence level is high. The slow change in fluorescence persists in PMSH₂-treated cells; in some cases it is enhanced. This may suggest that cyclic electron flow induces S to M rise but this does not seem to be the case, as PMSH₂ quenches S to M rise in the presence of DCMU. It is possible that the PMSH₂-induced slow fluorescence rise in *Anacystis* is linked to some sort of non-cyclic electron flow mediated via PS I.

DISCUSSION

Table 1 summarizes some of the general effects of various treatments on the different phases of the fluorescence transients. The observations presented earlier and listed in table 1 amply illustrate some of the difficulties encountered in explaining the slow changes in Chl a fluorescence. This is because these slow changes reflect many complex physiological processes associated with photosynthesis. In spite of these complexities, however, some general trends seem to be clear. Thus, several physical and physiological processes seem likely to be associated with the different phases of the transient in intact cells; these are listed in Table 2.

The P to S Decay

There is ample evidence (see I) that the P to S decline is not linked to the redox state of Q (also see Table I, items 4 and 5). The effect of preillumination (Figs 6-8) on the P to S decline clearly indicates that the latter is not directly linked to the redox state of Q. However, P to S decay is somewhat dependent on the flow of electrons between the two photosystems as DCMU and o-phenanthroline completely suppress it (Table 1, 6a). The inhibitor NH₂OH at high concentrations, but not at low concentrations, suppresses the P to S decline (39). The absence of P to S decline under anaerobiosis (Table 1; see also 4, 9, 18, 40, 41) indicates that the P to S decline may also be susceptible to quenching by molecular oxygen (17).

The P to S decline is extremely vulnerable to uncouplers (Table 1, 8a),

TABLE 1

A Summary of the Effects of various Treatments on Chlorophyll Fluorescence Transients in Algae

		Induction phase				
Treatment	Algae used	D to P	P to S	S to M	M to T	References
(1) Anaerobiosis (2) CO ₂ depletion	Chlorella Chlorella	Pronounced Present	Absent Present	Absent	Absent	4, 9, 18, 33, 41 16, 49
	Chlamydomonas	Enhanced	Enhanced			
(3) pH	Chlorella	Remained unaltered between pH 6 to 8	No Significant change	No chage	Abolished	16, 52
(4) Low Temperature (5°C)	Porphyridium	Slow rise	Slow decline	<u> </u>	**************************************	33, 42, 44, this paper
(5) Preillumination by: (a) PS I light (b) PS II light	Chlorella Porphyridium —do—	(a) Delayed low P level (b) Fast rise	Slow decline Fast decline	-		PS I: 4, 9, 34; PS II: 34, this paper
(6) System I light	Chlorella	Quenching at P		Less quenching	More quenching	11, 35
(7) Inhibitors of electron flow a) DCMU, and o-phenanthroline	Several	(a) Very fast rise to maximal level	Absent	Absent in Chlorella & Porphyridium but present in Anacystis	Absent	DCMU: 25, 28, 39, 58

TABLE 1 (Contd.)

1	2	3	4	5	6	7
b) NH ₂ OH		b) DP rise present	Absent	Slightly suppressed or not at all (Anacystis)	Absent	NH ₂ OH: 39 this paper
(8) Uncouplers of photophosphory-lation a) CCCP/FCCP	Chlorella	a) Lowers P; slow rise	Suppressed	Enhanced in Chlorella but not in CO ₂ -depleted Anacystis	Abolished	24, 25, 28, 46 this paper.
b) NH₄Cl	Chlorella	b) No change	No change	Stimulated in CO ₂ -depleted Chlorella	Absent	16, 48
(9) Electron Transport Carriers	Chlorella Porphyridium					
a) MV		a) Low P level; only OI rise	Absent	Suppressed	Absent	4, 9, this paper
b) PBQ	Chlorella	b) Low	Insignificant; absent	Slow rise	Absent	44, this paper
c) PMSH ₂	Anacystis	c) High P level	Absent	Persists	Absent	17, 24 this paper.
(10) High salt (KCl, MgCl ₂ or K acetate)	Chlorella	Quenching at P	-	Less quenching	More quenching	48.

TABLE 2

Possible Relationship of various Physical and Physiological Process with different Phases of Chlorophyl Fluorescence Transient

Induction Phase		Process linked to the transient phase	Reference	
(1)	O-I rise	Primary photochemical reaction at the reaction centre II; redox state governs the fluorescence yield	2,6.	
(2)	I-D decline	Interaction of Q with pool of acceptor A	4, 9.	
(3)	D to P rise	"Traffic jam" due to a bottleneck on the acceptor side of pigment system I; accumulation of reduced carriers	4, 9.	
(4)	P to S decline	Lowring of yield due to changes in "phosphate potential" or proton gradient or membrane potential leading to structural changes facilitating decrease in fluorescence yield; seems to be linked to loss of Mg ²⁺ ions; may be affected by molecular oxygen	17, 20.	
(5)	S to M rise	Independent of redox state of Q; regulated by structural changes of the membrane facilitating increase in fluorescence yield; may be linked to export of energy and uptake of materials between chloroplasts and their surrounding cellular constituents	28, 35, 50.	
(6)	M to T decline	Indirectly linked to synthesis and utilization of ATP in CO ₂ fixation	8, 2 5. 46.	

which indicates that this slow decline is somehow regulated by phosphorylation. One would expect that a cyclic electron flow supported photophosphorylation to be stimulated by anaerobiosis and DCMU poisoning. However, under these conditions no P to S decline is observed, at least, in *Chlorella* (46) and *Porphyridium* (8) which indicates that the P to S decline may be linked not to a cyclic electron flow, but to non-cyclic photophosphorylation in these algae. Experimental support in favour of this hypothesis may be taken from the results of Krause (19) who has shown that the slow P to S decline in intact chloroplasts is suppressed by uncouplers

of phosphorylation. Krause (20) and Barber et al. (21) have further demonstrated that the P to S decline is linked to the loss of Mg^{2+} ions from the grana stacks to the stroma region. Krause (20) concluded (also see 1) that a "phosphate potential" is responsible for a structural alteration of the membrane leading to the P to S decline. Although Krause's experiments indicate that loss of Mg^{2+} like cations may be counterbalanced with the exchange of protons, it is not certain if the loss of Mg^{2+} by grana into the stroma region is compatible with the H^+ uptake. Hind et al. (51) have shown that the uncoupler CCCP, but not NH^+_4 ions, inhibits Mg^{2+} efflux to the medium in isolated chloroplasts which lends support to the hypothesis (see Table 1, 8b) that loss of Mg^{2+} is linked to the P to S decline. In the blue green algae there is no clear demarcation of stroma and grana regions but these algae also exhibit the P to S decline (28). We do not yet know if there is any compartmentation in the lamellar regions of the blue-green algae.

The hypothesis of Krause (19) for the P to S decline differs from that of Papageorgiou (17) in that the latter proposed collisional-quenching of Chl a fluorescence by molecular oxygen. Papageorgiou (17) suggests that acidification of the inside of the thylakoid membranes makes the pigment bed more susceptible to quenching by O₂. This hypothesis is supported by the observation that PMSH₂ induced lowering of Chl a fluorescence in DCMU-treated isolated chloroplasts is suppressed at low O₂ tension (43). A difficulty in this hypothesis is the fact that during the P to S decline phase, fluorescence and oxygen evolution are complementary. This relationship would not be retained if substantial amount of O2 would be needed to quench Chl a excitation. Also, under anaerobiosis the O_2 evolution is not insignificant although the P to S decline is totally arrested (Table 1, 1). Furthermore, it has been shown that during the P to S decline the ratio of System I to System II emission is altered. No such alteration is seen in PMS-induced lowering of Chl a in isolated chloroplasts (22, 23); this indicates that the PMSinduced lowering in isolated chloroplasts may not be quite analogous to the P to S decline in algae. In spite of these objections the suggestion that the P to S decay may be partly due to quenching by molecular O₂ remains feasible.

Slovacek and Bannister (16) suggested that the P to S decline reflects an activation of some dark enzymatic step in the electron transport chain. Such activation would accelerate electron flow and reactivate PS II. However, this proposition cannot easily explain the initial observation of Duysens and Sweers (10) and the data shown in Fig. 9C that a dark period is necessary to restore the high level fluorescence. Thus, it appears that the most attractive theory for the P to S

decline would involve the postulation of a build-up of a "phosphorylation potential" due to non-cyclic electron flow: the fluorescence decline must occur because this potential creates structural changes on the membrane which lead to an increase in the rate constant of internal conversion and / or transfer to weakly fluorescent Chl a molecules.

The S to M rise

The S to M rise part of the transient seems to be more independent of the redox state of Q than the P to S decline phase (Table 1, 6, 7). This is evident from the facts that (a) the S to M rise is resistant to DCMU and o-phenanthroline (Table 1, 7) in Anacystis (50, 52) and (b) a development of the fast fluorescence transient, which is associated with redox reactions of the electron transport chain (see 1) is not a prerequisite (see Fig. 14) for the development of the S to M rise (Table 1, 9b & c). The S to M transient is not much affected by the pH of the surrounding medium (Table 1, 3) (52); also, the effect of system I light to quench the fluorescence evoked by PS II light is comparatively less in this phase than in the M to T phase (Table 1, 6) (35).

Uncouplers like CCCP and FCCP have somewhat different effects (Table 1, 8a) on the S to M rise phase in *Chlorella* and in *Anacystis*. In *Chlorella*, although CCCP and FCCP suppress both the P to S and the M to T declines, they do not affect S to M rise as shown in Fig. 15 (also see 46). In *Anacystis*, the S to M transient is suppressed at a low concentration of FCCP (25, 50). FCCP at high concentrations and salicylanalides at fairly low concentrations suppress the S to M rise in DCMU-treated *Anacystis* (50). We have also shown that the S to M rise is associated with structural alteration in this alga as measured by changes in optical density at 540 nm. In spite of the differential behavior of the S to M phase of the transient to DCMU (25, 28, 46, 50) (also see Table 1, 7a, b) and FCCP (Table 1, 8a) in *Chlorella* and *Anacystis*, it is reasonable to assume that some structural changes are associated with this S to M change in *Chlorella* as we have shown in the case of *Anacystis*; in both cases the fluorescence yield changes are not linked to Q.

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Slovacek and Bannister (49) have shown, as noted earlier, that in CO₂-depleted cells of *Chlorella*, the S to M phase is suppressed; however, the addition of NH₄Cl to CO₂-depleted samples revives a large S to M rise and also increases the steady state fluorescence level. These authors argue that such large fluorescence yield changes observed by the addition of NH₄Cl cannot be accommodated by the

change in distribution of quanta between the two photosystems, i.e., the so called α -change hypothesis, but can be explained by activation of PS II units.

The stimulation of S to M rise in Chlorella in the presence of FCCP (Fig. 15) is somewhat similar to the effect of NH₄Cl on the CO₂-depleted Chlorella. Also the S to M rise in *Chlorella* is practically unaffected by pH of the medium (52). Slovacek and Bannister (49) have suggested that there is possibly no large proton gradient in CO₂-supplemented whole cells of Chlorella during steady state photosynthesis as fluorescence level remains unaltered with or without NH₄Cl. Thus the S to M rise may not be linked to any proton gradient, and the uncouplers must be acting on the structure of chloroplasts in some indirect way. The stimulation of S to M rise in the presence of FCCP in Chlorella and the inhibition of the S to M rise in DCMU-treated Anacystis by high concentrations of FCCP and salicylanalides indicate that the S to M rise may be related to structural organization of the thylakoid membrane. It seems to us that the S to M rise may be similar to the cation-stimulated (23, 53, 54) fluorescence rise in isolated chloroplasts which is not directly linked to the energy conservation process. However, in intact cells, the export of energy from the chloroplast to the cytoplasm or the cell may be accompanied by some slow active import of Mg²⁺ from the cytoplasm (55). Such active import of materials like sugars (e.g., glucose) or anion would be expected to be affected indirectly by uncouplers. Although some of the experimental data cannot be adequately explained by the above suggestion, it seems very clear that the S to M rise originates from structural alteration that facilitates transformation of photostates in favour of the so-called state I (see 1, 14, 17).

The M to T decline

The M to T decline phase has been shown to be very susceptible to growth conditions, uncouplers of phosphorylation (25), CO₂ content (16) and energy transfer inhibitors (25). The slow M to T decline in *Porphyridium* is also suppressed by an energy transfer inhibitor like Phlorizin (8, 25). The slow M to T decline thus, seems to reflect the slow processes of ATP synthesis and utilization during photosynthesis. It is clear from our past work and that of others that the major slow fluorescence yield changes in intact cells are not linked to the photochemical reaction but associated with the energy conservation and other physiological processes: they are linked to phosphorylation in the form of proton gradient or membrane potential and are associated with the structural alteration of the membranes linked with photo-transformation of states.

Concluding remarks

The P to S decay is retarded by PS I preillumination and is accelerated by PS II preillumination, while the D to P rise exhibits opposite effects under these conditions. The P to S decay is slowed down at 5°C as compared to 23°C while the D to P rise does not show significant change at these two temperatures. The P to S decay is abolished by inhibitors of electron flow (e.g. DCMU, o-phenanthroline), inhibitors of O₂ evolution (high concentrations of NH₂OH) and anaerobiosis. Under these conditions the fluorescence level usually remains high and does not show P to S decay. Also, no P to S decline is observed in the presence of the terminal PSI acceptor methyl viologen. Methyl viologen keeps the Chl a fluorescence at a low steady S level-mostly due to the oxidation of carriers in the electron transport chain. The above results suggest that a normal photosynthetic O, evolution may be necessary for the development of the P to S decline although it does not reflect oxidation-reduction of Q. The P to S decline is extremely sensitive to uncouplers like CCCP and FCCP. This suggests that the P to S decline may be linked to "energy conservation" process coupled to non-cyclic electron flow. However, the results obtained with methyl viologen (MV) and NH₂OH cannot be easily explained on the basis of this hypothesis. One must assume that in the presence of NH₂OH cells lose their ability to phosphory late. and in the presence of MV, the oxidation level of Q and the build up of "phosphate potential" keep the yield of fluorescence at a minimum level.

The slow S to M rise is also not associated with the non-cyclic electron flow. The S to M rise is insensitive to DCMU. Although MV arrests the S to M rise. p-benzoquinone (PBQ), which suppresses the P to S decline, has no or slightly stimulatory effect. The S to M rise is not as sensitive to uncouplers of phosphorylation as the P to S decay is. In Chlorella, the S to M rise is resistant to low concentrations of CCCP and FCCP; at high concentrations, they stimulate the S to M rise. In Anacystis, high concentrations of FCCP inhibit the S to M rise in presence of DCMU. Furthermore, the S to M rise is resistant to NH₂OH in DCMU-treated Anacystis. PMS suppresses the S to M fluorescence rise in DCMU-treated cells of Anacystis; it does not affect the S to M rise in untreated cells, although it abolishes the P to S decline. In untreated cells, PMS seems to act as a Hill oxidant like PBQ. The above results are complex and somewhat variable with organisms. yet some generalizations can be made: (i) The S to M rise does not need sustained non-cyclic electron flow: (ii) it is partially resistant to uncouplers of phosphoryltion and (iii) it exhibits large alterations in the emission characteristics of Chl a fluorescence (8, 15). These results, together with our earlier observation that the

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S to M rise is arrested by glutaraldehyde fixation, indicate that this slow change in fluorescence yield may be analogous to the cation-induced fluorescence yield changes seen in isolated broken DCMU-treated chloroplasts (23, 54). Thus the S to M rise seems to reflect phototransformation of 'states'. We assume that the S to M changes in whole cells may be linked to export of energy from the chloroplasts and import of ions (or sugars) to the chloroplasts.

The ultimate causes of fluorescence yield changes must be in changes in either (a) rate of photochemical reactions or (b) rate constant of excitation energy transfer among strongly fluorescent and weakly fluorescent Chl a molecules, or (c) rate of internal conversion. Both the P to S and the S to M changes seem to occur partially through associated microstructural changes of membrane matrix that lead to changes in (b) or (c).

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REFERENCES

- 1. GOVINDJEE & PAPAGEORGIOU, G., In Photophysiology Vol. 6, 1-46. A. C. Geise, editor. Academic Press, New York (1971).
- 2. JOLIOT, P., Biochim, Biophys. Acta, 102, 135 (1965).
- 3. KOK, B., FORBUSH, B. & MCGLOIN, M., Photochem, Photobiol., 11, 457 (1970).
- 4. MUNDAY, J. & GOVINDJEE, Biophys. J., 9, 1 (1969).
- 5. Delsome, R., Ph. D. Thesis, University of Paris (1967).
- 6. JOLIOT, P., Photochem. Photobiol., 8, 451 (1968),
- 7. BANNISTER, T. & RICE, G., Biochim. Biophys. Acta, 162, 555 (1968).
- 8. MOHANTY, P., PAPAGEORGIOU, G. & GOVINDJEE, Photochem. Photobiol., 14, 667 (1971).
- 9. MUNDAY, J. & GOVINDJEE, Biophys J., 9, 22 (1969).
- DUYSENS, L. N. M. & SWEERS, H., In Studies in Microalgae and Photosynthetic Bacteria pp. 352-372, Japanese Society of Plant Physiology, editors, Tokyo (1963).
- DUYSENS, L. N. M., Conference on Photosynthetic Unit, Abstr. A4, Gatlinburg, Tenn., U.S A. (1970).
- 12. MURATA, N., Biochim, Biophys, Acta, 172, 142 (1969).
- 13. BONAVENTURA, C. & MYERS. J., Biochim. Biophys. Acta, 189, 289 (1969).
- 14. MYERS, J., Annu. Rev. Plant Physiol., 22, 289 (1971).
- 15. Papageorgiou, G. & Govindjee, Biophys J., 7, 375 (1967).
- 16. SLOVACEK, R. & BANNISTER, T., Biochim. Biophys. Acta, 292, 729 (1973).
- 17. PAPAGEORGIOU, G., In Bioenergetics of Photosynthesis, pp. 319-369, Govindjee, editor. Academic Press, New York (1975).
- 18. KAUTSKY, H., APPEL, W. & AMMANN, H., Biochem. Z., 232, 277 (1960).
- 19. KRAUSE, G., Biochim. Biophys. Acta, 292, 715 (1973).
- 20. KRAUSE, G., Biochim. Biophys. Acta, 333, 301 (1974).
- 21. BARBER, J., TELFER, A. & NICOLSON, J., Abstracts, 3rd International Congress on Photosynthesis, Sept. 2-6, Rehovot, Israel, p. 6 (1974).

- 22. MURATA. N. & SUGAHARA, K., Biochim. Biophys. Acta, 205, 371 (1970).
- 23. MOHANTY, P., BRAUN, B. Z., & GOVINDJEE, Biochim. Biophys. Acta. 292, 459 (1973).
- 24. MOHANTY, P. & GOVINDJEE, Photosynthetica, 7, 146 (1973).
- 25. PAPAGEORGIOU, G. & GOVINDJEE, Biophysic. J., 8, 1396 (1968).
- 26. GOVINDJEE & RABINOWITCH, E., Biophys. J., 1, 73 (1960).
- 27. SHIMONY, C., SPENCER, J. & GOVINDJEE, Photosynthetica, 1, 141 (1967).
- 28. MOHANTY, P. & GOVINDJEE, Plant & Cell Physiol., 14, 611 (1973).
- 29. MUNDAY, J., Ph. D. thesis, University of Illinois (1968).
- 30. LAVOREL, J., Editions du CNRS, 119, 161 (1963).
- 31. GOVINDJEE, PAPAGEORGIOU, G. & Munday, J., Brookhaven Symposium in Biology, 19, 134 (1966).
- 32. RABINOWITCH, E. I.. Photosynthesis and Related Processes, Vol. 2, pp. 1390-91, Wiley Interscience, New York (1956).
- 33. Franck, U. F., Hoffman, F., Arenz, H. & Schreiber, U, Bur. Bunsenges Phys. Chem., 73, 87 (1969).
- 34. VREDENBERG, W., Biochim. Biophys. Acta, 189, 129 (1969).
- 35. MOHANTY, P., MUNDAY, J. C. & GOVINDJEE, Biochim. Biophys. Acta, 223, 198 (1970).
- 36. HEATH, R. L., Biophys. J., 10, 1173 (1970).
- 37. DUYSENS, L. N. M., Biophysic. J., 12, 858 (1972).
- 38. BENNOUN, P. & JOLIOT, A., Biochim. Biophys. Acta, 189, 85 (1969).
- 39. MOHANTY, P., MAR, T. & GOVINDJEE, Biochim, Biophys. Acta, 253, 213 (1971).
- 40. Franck, U. F. & Hoffman, N., Progress in Photosynthesis Research, 2, 829 (1969).
- 41. SCHREIBER, U., BAUER, R. & FRANCK, U. F., Z. Naturforsch. 26b, 1195 (1971).
- 42. SIMONIS, W. & URBACH, W., Annu. Rev. Plant Physiol., 24, 89 (1973).
- 43. PAPAGEORGIOU, G., ISSAKIDOU, G. & ARGOUDELIS, G., FEBS Letters 25, 139 (1972).
- 44. MOHANTY. P., Ph. D. thesis, University of Illinois (1972).
- 45. GOVINDJEE, R., Ph. D. thesis, University of Illinois (1961).
- 46. PAPAGEORGIOU, G. & GOVINDJEE, Biophys. J., 8, 1299 (1968).
- 47. AMESZ, J. & FORK, D., Biochim, Biophys. Acta, 143, 97 (1967).
- 48. MOHANTY, P., GOVINDJEE & WYDRZYNSKI, T., Plant & Cell Physiol., 15, 213 (1974).
- 49. SLOVACEK, R. & BANNISTER, T., Biochim. Biophys. Acta, 325, 114 (1973).
- 50. MOHANTY, P. & GOVINDJEE, Biochim. Biophys. Acta, 305, 95 (1973).
- 51. HIND, G., NAKATANI, H. Y. & IZAWA, S. I., Proc. Natl. Acad. Sci. U.S.A., 71, 1484 (1974).
- 52. PAPAGEORGIOU, G. & GOVINDJEE, Biochim, Biophys. Acta, 234, 428 (1971).
- 53. HOMANN, P., Plant Physiol., 44, 932 (1969).
- 54. MURATA, N., Biochim. Biophys. Acta, 189, 171 (1969).
- 55. LIN. D. C. & NOBEL, P., Archiv. Biochem. Biophys., 145, 622 (1971).